

CHEM F213: Physical Chemistry-II

Date: October 09, 2017

Time: 90 minutes

Mid-semester Test (Closed Book)

Max. Total Marks 60

General Instructions:

1. Write your name, ID no., Course No., Course Title, Date and Day legibly, correctly and completely on the main answer-sheet as well as on the supplement(s) (if any) used. **Incomplete or wrong information will result in deduction of up to 5 marks.**

2. All the questions are compulsory. You may solve the questions in any order. However, solve all the sub-questions of a question before going to new question.

3. Only scientific and non-programmable calculators may be used for numerical calculations. Use of calculators with operating systems is strictly prohibited. Use of mobile phones, iPhones, pagers, are strictly prohibited.

Q. 1. Consider a particle confined in a rectangular box such that $U(x,y)=0$ for $0 \leq x \leq a$; $0 \leq y \leq 2a$ and $U(x,y) = \infty$ otherwise.

(a) Write expression for normalized wavefunction in terms of the quantum numbers n_x and n_y . [2]

(b) Identify the lowest four energy states (write the quantum numbers only) and calculate the separation between the first and the fourth energy states in terms of $h^2/8ma^2$. [5]

(c) Calculate the probability that the particle in the state corresponding to $n_x=n_y=2$ is found in the region: $0 \leq x \leq a$, $0 \leq y \leq a/2$. [3]

Q. 2. Consider a particle moving with uniform circular motion in an orbit with a fixed radius. The stationary states of the system are described by Hamiltonian eigenfunctions:

$\Phi_m(\phi) = \frac{1}{\sqrt{2\pi}} e^{-i\mathcal{E}[\phi - (m\hbar t/2\mu\rho^2)]}$; $m=0, \pm 1, \pm 2, \dots$ where, μ is the mass of the particle and, ρ its distance from the centre of the orbit. In 2D-polar coordinate system, the operator ∇ takes the form:

$$\nabla = \frac{1}{\rho} \left(\frac{\partial}{\partial \rho} \rho + \frac{\partial}{\partial \phi} \right)$$

(a) Obtain the expression for probability current. [7]

(b) Calculate the probability current corresponding to $m=0, +1, -1$; respectively. [3]

Q. 3. The electron-spin functions, α and β are degenerate eigenfunctions of spin-angular momentum squared operator σ^2 with eigen value $0.75\hbar^2$. These functions also obey eigenvalue equation of the operator σ_z

(z-component of spin-angular momentum) and the corresponding eigenvalues are $0.5\hbar$ and $-0.5\hbar$, respectively.

(a) Define the ladder operators, σ_{\pm} . [2]

(b) Using known concepts about spin angular momentum of electron, guess the effect of action of the raising and the lowering operators on these eigenfunctions [4]

(c) Check whether the following functions: $(\sigma_+ \beta \pm i \sigma_- \alpha)$ are eigenfunction of (i) σ_z ; and (ii) σ^2 . [4]

Q. 4 Given $F = \{f_1, f_2, f_3, f_4\}$ is a complete orthonormal set of eigen-functions of hermitian operator, **A** with eigen-values $a_1=5, a_2=a_3=7, a_4=9$. Another complete orthonormal set $G = \{g_1, g_2, g_3, g_4\}$ is related to the set F as follows: $g_1 = 0.7071(f_1 + f_4)$; $g_2 = 0.7071(f_2 + f_3)$; $g_3 = 0.7071(f_2 - f_3)$; $g_4 = 0.7071(f_1 - f_4)$; The functions in G happen to be eigen-functions of another hermitian operator **B**, with eigen-values $b_1=6; b_2=b_3=8, b_4=10$. The sets F and G form two distinct complete sets of degenerate eigenfunctions of yet another hermitian operator **D** with eigenvalue $d=25$.

(a) Write the commutation relationships between these operators. [2]

(b) If measurement is made with operator A and value 9 is found, what is the state of the system immediately after the measurement? [2]

(c) Immediately after the measurement in (b), if measurement with operator B is done, what is the probability of getting the value 6? [2]

(d) If measurement with operator A is done and value 7 is found, measurement with operator B is done immediately after that, what is the probability that value 8 will be found? [2]

(e) If the system is prepared to be in the normalized state: $\psi = 3^{-1/2}f_1 + 0.5(f_2 + f_3) + c_4f_4$; (I) Find the value of the positive constant c_4 . (II) What will be the average value of operator **D** in this state? [2]

Q. 5. (a) Consider the following orbital of a hydrogen atom:

$$\Psi(r,\theta,\phi) = (3/\pi a_0^3)^{1/2} (r^3/187500 a_0^3) (20 - r/a_0) e^{-r/5a_0} \sin(\theta) \sin(2\theta) \cos(2\phi)$$

(i) Identify the number of radial node(s) and locate the same. [1]

(ii) Identify the number of angular nodes. [1]

(iii) Identify the orbital. [2]

(iv) Is it an eigenfunction of L_z operator? If yes, what is the eigenvalue? If no, what is $|m_l|$? [2]

(b) Hermite polynomials defined by $H_n(y) = (-1)^n e^{y^2} d^n (e^{-y^2})/dy^n$, obey the Hermite's differential equation:

$$H''(y) - 2yH'(y) + (\lambda - 1)H(y) = 0. \text{ Prove that } yH_n(y) = (1/2)H_{n+1}(y) + nH_{n-1}(y) \quad [4]$$

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