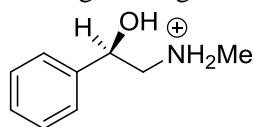
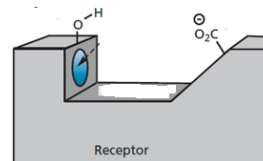


Q1. (a) Explain the distinction between a binding site and a binding region. 3

(b) Schematically explain the binding of a chemical messenger to a protein receptor. Illustrate the binding of the given neurotransmitter to a binding site of receptor 6

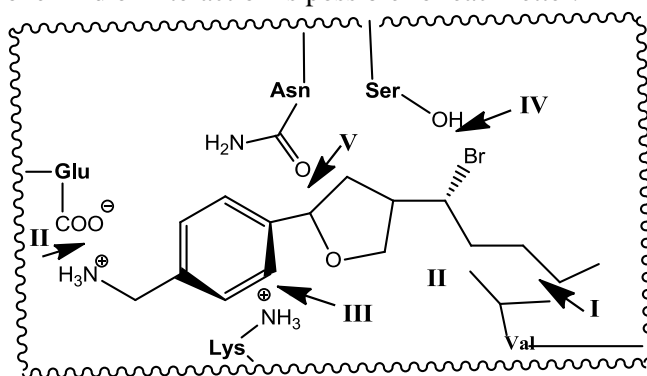


Neurotransmitter



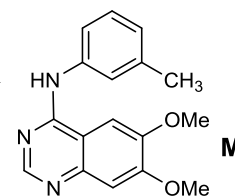
(c) Briefly explain, (i) Dipole-dipole and ion-dipole interactions 4 (ii) Prodrugs

(d) Indicate what drug-receptor interactions are involved at every arrow shown in the figure below. More than one kind of interaction is possible for each letter. 5



I _____
 II _____
 III _____
 IV _____
 V _____

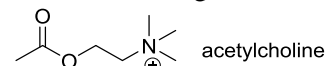
(e) 4-Anilinoquinazoline **M** had useful *in vitro* activity but its *in vivo* activity was hampered as it gets rapidly metabolized by cytochrome P450 to produce two metabolites. Identify the structures of metabolites and modified structure which is resistance to metabolism. 4



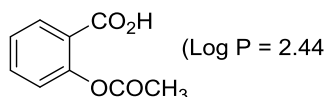
Q2 (a) Identify the intermolecular/intramolecular that are possible for the side chain of Phenylalanine and Serine 3

(b) What is the one letter code for the polypeptide, Glu-Leu-Pro-Asp-Gly-Thr 3

(c) Acetylcholine is the substrate for the enzyme acetylcholinesterase. Suggest what sort of binding interactions could be involved in holding acetylcholine to the active site 4



Q3. (a) Describe Lipinski's rule of five. Verify the Lipinski's rule of five for the following drug molecule. 4



(b) A lead compound has a mono-substituted aromatic ring present as part of its structure. An analogue was synthesized containing a para chloro substituent which had approximately the same activity. It was decided to synthesize an analogue bearing a methyl group at the para position. This showed increased activity. What analogue would you prepare next and why? 4

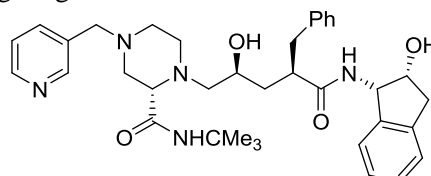
(c) Calculate the log P value for *p*-methylphenol (Values of π for a range of substituents) 3

Group	CH ₃	<i>t</i> -Bu	OH	OCH ₃	CF ₃	Cl	Br	F
π (aliphatic substituents)	0.50	1.68	-1.16	0.47	1.07	0.39	0.60	-0.17
π (aromatic substituents)	0.52	1.68	-0.67	-0.02	1.16	0.71	0.86	0.14

(d) Phenols are antiseptics. Hansch analysis carried out on a series of phenols with the general structure **A** (p -RC₆H₄OH) yielded the Hansch equation $\log 1/C = 1.5 \Pi - 0.2 \sigma + 2.3$ ($n = 23$, $s = 0.13$, $r = 0.87$) 5

What is (i) the significance of the terms n , s and r , (ii) the relative significance of the lipophilicity and electronic distribution of a phenol of type **A** on its activity and (iii) the effect of replacing the R group of **A** by a more polar group?

(e) Write the metabolites of antiviral agent given below 4



Q4. (a) Write the target and mechanism of action for chlormethine [MeN(CH₂CH₂Cl)₂]. Why chlormethine is too reactive to survive the oral route? Design an analogue of chlormethine with reduced side effects. 5

(b) Lineweaver-Burk plots are extremely useful in determining the nature of inhibition. Draw Lineweaver-Burk plots with and without competitive inhibitors present 3