

Birla Institute of Technology & Science, Pilani

Semester I (Session 2016-17)

Mid Semester Examination (CB)

STATISTICAL MECHANICS

Max. Time: 1.5 hrs

Max. Marks : 60

Q1. Suppose a thermal system of N independent particles in d -dimensions at a fixed temperature T . The single particle Hamiltonian of the system is given as, $h = A|\vec{p}|^m + B|\vec{q}|^n$. Compute the (ensemble) average energy $\langle \epsilon \rangle$ of each particle by applying equipartition theorem, $\langle x_i \frac{\partial h}{\partial x_j} \rangle = \delta_{ij} KT$. Where x_i denotes the coordinates q & p . [10]

Q2. Consider a system of N (distinguishable) particles, where the possible energy values for a single particle are $0, \epsilon_1, \dots, \epsilon_n$. The energy value ϵ_a is g_a - fold degenerate, and the ground state is non-degenerate. (i) If $p_0(T)$ is the probability for a single particle to be in the ground state, determine $p_0(T \rightarrow \infty)$ and $p_0(T \rightarrow 0)$. (ii) Calculate average energy $\langle \epsilon \rangle$ per particle at $T \rightarrow \infty$. [5 + 3]

Q3. N diatomic (distinguishable) molecules are stuck on a metal surface. Each molecule can either lie flat on the surface, in which case it must be aligned to one of two directions, $+x$ and $+y$, or it can stand up along the $+z$ direction. There is an energy cost of ϵ (> 0) associated with a molecule standing up, and zero energy for molecules lying flat along x or y directions. Suppose N_z number of molecules stand up along z -direction at a temperature T . Use microcanonical ensemble to determine the ratio of $\frac{N_z}{N}$ for which entropy of the system becomes maximum. (Use Stirling approximation, $N! = N \ln N - N$.) [12]

Q4. Consider a rod-shaped diatomic molecule with moment of inertia I , and a (electric) dipole moment μ . The contribution of the rotational degrees of freedom to the Hamiltonian is given by, $H = \frac{p_\theta^2}{2I} + \frac{p_\phi^2}{2I \sin^2 \theta} - \mu E \cos \theta$. Where E is an external electric field. θ & ϕ are the usual polar and azimuthal angles, respectively, while p_θ & p_ϕ are their conjugate momenta. (a) Calculate the classical partition function $Q_1(T)$ of each dipole. (b) Obtain the average polarization, $P = \langle \mu \cos \theta \rangle$ of each dipole. (You can use, $\int_0^\infty e^{-cx^2} dx = \frac{\sqrt{\pi}}{2\sqrt{c}}$; $\int_0^\infty e^{-x} x^{n-1} dx = \Gamma(n)$; $\Gamma(1/2) = \sqrt{\pi}$.) [12 + 6]

Q5. N (non-interacting) hydrogen molecules H_2 absorbed on a (two-dimensional) flat surface are in thermal equilibrium at temperature T . The rotational motion of each molecule is confined to the plane of the surface. The quantum state of the planar rotation is specified by a single quantum number m which can take on the values $0, \pm 1, \pm 2, \pm 3, \dots, \pm \infty$. There is one quantum state for each allowed value of m . The energies of the rotational states are given by $\epsilon_m = (\hbar^2/2I)m^2 \equiv \epsilon m^2$, where I is a moment of inertia. Determine the rotational energy contribution to the heat capacity C of the gas at high temperature T , where $KT \gg \epsilon$. In this limit, *sum* can be replaced by *integration* on variable m . [12]

BEST OF LUCK