

1. A particular system obeys the equations $U = \frac{PV}{2}$ and $T = \left(\frac{DU^{3/2}}{VN^{1/2}}\right)^{1/2}$, where D is a positive constant.

- (a) Find the fundamental relation $S(U, V, N)$ for the system. (*Hint: You may have to use the fact that entropy is an extensive quantity to fix the fundamental relation completely, except possibly for a constant coming via integration.*)
- (b) Find chemical potential of the system as a function of T, V and N .
- (c) Find the Helmholtz free energy, $A(T, V, N)$.

[6 + 4 + 4]

2. Consider an ideal gas present in a cylindrical container of radius R and height, H ($H \ll$ radius of the earth) on the surface of the earth. The total number of particles in the gas is N and mass of each of the particles is m . The total energy of the gas is E .

- (a) Find the equilibrium distribution function, $f(\vec{p}, \vec{r})$, in the μ -space of the system. (You may leave the answer in terms of the Lagrange multiplier, β , appearing in the MB - distribution function.)
- (b) What is the average speed of the particles in the cylinder?
- (c) Is the average speed of the particles in the upper half of the cylinder less than, equal to or more than the value found above? Justify.
- (d) Comment about the density distribution of the gas inside the cylinder for the cases when: (i) $\frac{E}{N} \ll mgH$ and (ii) $\frac{E}{N} \gg mgH$.

[6 + 4 + 2 + 2]

3. The relation between momentum and energy for an extreme relativistic particle is $\epsilon = pc$. Consider an ideal gas consisting of N extreme relativistic particles of mass m and total energy E confined to a volume, V .

- (a) Find the fundamental relation, $S(E, V, N)$ for the gas.
- (b) Find the pressure of the gas as a function of energy density.

4. Let $\eta(\theta_1, \theta_2)$ be the efficiency of a Carnot engine working between two thermal reservoirs (with $\theta_1 > \theta_2$). Show that $\eta(\theta_1, \theta_2) = \eta(\theta_1, \theta_3)$ if and only if $\theta_2 = \theta_3$. You may not use the expression for η in terms of the absolute temperature of the reservoirs as that scale is set up making use of the above result. (*Hint: Look for violation of one of the statements of the second law of thermodynamics.*) [6]

Useful Formulae:

- $S = k_B \text{Log}(\Omega)$
- $\int \dots \int_{0 \leq \sum_{i=1}^M r_i \leq R} \prod_{i=1}^M 4\pi r_i^2 dr_i = \frac{(8\pi R^3)^M}{(3M)!}$
- $\int_{-\infty}^{\infty} e^{-\alpha x^2} = \sqrt{\frac{\pi}{\alpha}}$
- $dU = TdS - PdV + \mu dN$
- Helmholtz free energy, $A = U - TS$
- Gibbs, free energy, $G = A + PV$
- Maxwell - Boltzmann distribution: $C e^{-\beta(\frac{p^2}{2m} + V(\vec{r}))}$