# BIRLA INSTITUTE OF TECHNOLOGY \& SCIENCE, PILANI 

II SEMESTER 2021-22

Comprehensive Examination

Soft Condensed Matter Physics (PHY F416))
Date: 20-05-2022
Max Time: 120 min
Max Marks: 80

## IMPORTANT:

Mere deriving an expression is not a qualification of getting full marks. You have to explain the terms and equations that are appearing in the process of solving the problem.

1. (10 marks) Consider a solid with a simple cubic structure, for which the interatomic potential $U(r)$ is given as,

$$
U(r)=\frac{A}{r^{n}}-\frac{B}{r^{m}}
$$

Having first derived expressions for the equilibrium separation $a$ and the energy at equilibrium $-\epsilon$, derive an expression for the Young's modulus in terms of the bond energy $\epsilon$ and the equilibrium separation $a$. Assume that when a tensile stress is applied, the interatomic separations in directions perpendicular to the stress remain unchanged.
2. ( $\mathbf{1 5}$ marks) The phase behaviour of a certain liquid mixture can be described by the regular solution model. Consider a solution of two particles. By taking proper assumptions, prove that the free energy density of the mixing of solution is given by:

$$
f(\phi)=\frac{k_{B} T}{v_{c}}[\phi \ln \phi+(1-\phi) \ln (1-\phi)+\chi \phi(1-\phi)]
$$

where, $\phi=\frac{N_{p}}{N_{T}}, N_{T}=N_{p}+N_{s}$, and $v_{c}=\frac{V}{N_{T}}$ and $\chi$ is the interaction parameter.
3. (a) ( $\mathbf{1 0}$ marks) Show that the terminal velocity for sedimentation of colloidal spherical particles in a Newtonian fluid (water) is:

$$
v=\frac{2}{9} \frac{R^{2} \Delta \rho g}{\eta}
$$

where $\eta$ is the fluid's viscosity, $\Delta \rho$ is the difference in the fluid densities, and $R$ is the radius of the particle.
(b) ( 5 marks) Find out the density of the polystyrene bead of radius is $2 \mu \mathrm{~m}$. Given that the terminal velocity of the bead is $0.44 \mu \mathrm{~m} / \mathrm{s}$, the viscosity of water is $1.002 \times 10^{-3} \mathrm{~Pa}$-s, and the density of the bead is $1.05 \mathrm{~g} / \mathrm{cm}^{3}$. Take $\rho_{w}=1 \mathrm{~g} / \mathrm{cm}^{3}$.
4. (15 marks) Consider a polymer chain for which the probability distribution function of end-to-end distance is:

$$
P(\boldsymbol{r}, N)=\left(\frac{2 \pi N a^{2}}{3}\right) \exp \left(-\frac{3 \boldsymbol{r}^{2}}{2 N a^{2}}\right)
$$

where, $\boldsymbol{r}$ is the end-to-end distance, $N$ is the number of links each length $a$. We found that $\langle\boldsymbol{r}\rangle \sim N^{\nu}$ The ideal chain model suggests that $\nu=0.5$. Find out the critical exponent $\nu$, for real chain by considering the excluded volume effect.
5. (5 marks) An order parameter for a nematic phase can be described by the second order Legendre polynomial, $S=\frac{1}{2}\left\langle 3 \cos ^{2} \theta-1\right\rangle$. Solve this equation for an order parameter $S=1$ and $S=0$. Comment on the obtained results.
6. (5 marks) The free energy per molecule transforming from isotropic to nematic phase according to Maier-Saupe theory is:

$$
\Delta F=\frac{1}{2} u S^{2}+k_{B} T \int f(\theta) \ln [4 \pi f(\theta)] d \Omega
$$

where, $u$ is a parameter that expresses the strength of the favourable interaction between two neighbouring molecules, and $S$ is the order parameter. $f(\theta)$ is the distribution function. With the help of the above equation, we plot the free energy as a function of the order parameter (as shown in the figure). The five different curves are for the different values of the coupling parameter, $u$. With proper
 justification, explain which curve is for the highest value of $u$.
7. Suppose that the unit vector $\boldsymbol{u}$ is isotropically distributed on the sphere $|\boldsymbol{u}|=1$. Let $\langle\ldots\rangle_{0}$ be the average for this distribution

$$
\langle\ldots\rangle_{0}=\frac{1}{4 \pi} \int d \boldsymbol{u}
$$

Derive the following equation:
(a) (5 marks)

$$
\left\langle u_{z}^{2}\right\rangle_{0}=\frac{1}{3}, \quad\left\langle u_{z}^{4}\right\rangle_{0}=\frac{1}{5}
$$

(b) (5 marks)

$$
\left\langle u_{z}^{2 m}\right\rangle_{0}=\frac{1}{2 m+1}
$$

8. (5 marks) When we apply stress on the liquid crystal, it responds differently from the solids or liquid. Using a proper mathematical expression explain the kinds of distortions in liquid crystals under applied stress.

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II SEMESTER 2021-22
Comprehensive Examination
Soft Condensed Matter Physics (PHY F416))
Date: 20-05-2022
Max Time: 60 min
Max Marks: 40

## IMPORTANT:

Mere deriving an expression is not a qualification of getting full marks. You have to explain the terms and equations that are appearing in the process of solving the problem.

1. ( $\mathbf{1 5}$ marks) Find out the mean square displacement of the free Brownian particle using the following equations:

$$
\begin{aligned}
\left\langle v(t) v\left(t^{\prime}\right)\right\rangle & =\frac{k_{B} T}{m} \exp \left(-\frac{t-t^{\prime}}{\tau_{v}}\right) \\
\left\langle[x(t)-x(0)]^{2}\right\rangle & =2 \int_{0}^{t} d t \int_{0}^{t^{\prime}} d t_{2}\left\langle v\left(t^{\prime}\right) v(0)\right\rangle
\end{aligned}
$$

Show that in the region $|t| \gg \tau_{v}$, the mean square displacement is $2 D|t|$.
2. A solution is held in a cylinder sealed by two pistons and divided into two chambers by a semi-permeable membrane. Initially, the concentrations of the two chambers are the same at number density, $n_{0}$, and the chambers have the same volume $h A$, where $h$ is the height, and $A$ is the cross-section of the chamber. A weight $W$ is placed on top of the cylinder, causing the piston to move down.
(a) ( $\mathbf{1 5}$ marks) Ignoring the density of the solution and assuming that the
 solution is ideal, find out the displacement $x$ of the piston at equilibrium.
(b) ( 5 marks) If we consider the density of the solution $\rho$, find out the displacement $x$ of the piston.
3. ( 5 marks) Consider a colloid of charged spheres all of the radius $0.1 \mu \mathrm{~m}$ in an aqueous sodium chloride solution. Find out the Debye screening length for salt concentration of $10^{-5}, \& 10^{-2} \mathrm{~mol} / \mathrm{dm}^{3}$. Compare your results for these two concentrations.

